

Impedance Study on Non-Faradaic Process of Hierarchical Hollow Core-Mesoporous Shell Carbon

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Introduction

- The complexity of the porous structure affects electron conduction, mass-transport in the pores and chargetransfer¹.
- Controlled structure of hierarchical hollow core-mesoporous shell carbon (HCMS) as supports for electrochemical applications, provides enhanced performance^{2, 3}.
- > The objective of this study is to unveil the non-faradic and faradic units lithium based arrays



Fig. 3 (a) Nyquist plots, and (b) curves of imaginary part versus frequency for symmetrical HCMS/HCMS and AC/AC cells.

faradic process of HCMS carbon in lithium-based organic electrolyte by electrochemical impedance techniques.

Experimental

- 3D hierarchical hollow core-mesoporous shell carbon material (HCMS) was synthesized via replication of assembly of core-shell silica spheres².
- The electrolyte employed was 1 M LiPF₆ in a 1:1:1 (v:v:v) mixture of EC, DMC and EMC.
- > Symmetrical cells were used for impedance tests:







Ν

R_{pore-total}

Fig. 4 (a) Equivalent circuit model used for impedance fitting, and (b) transmission line model (TLM).

 Table 1
 Impedance
 parameters
 of
 symmetrical
 HCMS/HCMS

 and AC/AC cells
 AC/AC cells<

 R_{ec}/Ω

 τ_{ec}/s

 $R_{\rm b}/\Omega$

Results



Fig. 1 SEM images of (a) core-shell silica spheres, (b) silica/carbon composite after impregnation with carbon precursor and carbonization, and (c) as-prepared HCMS material after removal of silica template. TEM images of (d) silica spheres, (e) HCMS material, and (f) supercapacitor grade activated

HCMS/HCMS cell	1.6	9.3	1.56*10 ⁻⁵	3	84.3
AC/AC cell	1.4	15.2	3.16*10 -5	5	184.3

R_b: resistance of bulk electrolyte

 R_{ec} , τ_{ec} : resistance and time constant of electron conduction process N: number of TLM branches

 $R_{pore-total} = \sum R_{pore-n}$ (n=1, 2..., N), which reflects resistance of ions diffusion in pores

Conclusions

- > 3D hierarchical carbon framework with hollow core, mesoporous shell was synthesized.
- In lithium organic electrolyte, the Nyquist plot shows typical characteristic of porous electrode for HCMS carbon. One semicircle corresponds to electron conduction process; and an inclined -45° line followed by a nearly vertical line is observed to reflect ion diffusion inside pores.

carbon (AC).

Fig. 2 (a) N₂ sorption isothermals, and (b) DFT pore size distribution curves for HCMS and AC materials.

 Controlled bimodal HCMS structure has smaller resistance of ion diffusion in pores. Probably due to the unique hierarchical nanostructure facilitating effective ion transfer.
Extensive work in other electrolytes is still on-going.

References

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